

## Science Chemistry

### Unit 3: Petroleum: Breaking and Making Bonds

<b>Essential Understandings</b>	<ul style="list-style-type: none"> <li>▪ The physical world contains basic elements whose structure can be studied.</li> <li>▪ Matter is transformed in accordance with various chemical laws and principles.</li> <li>▪ Energy is a fundamental part of physical and chemical changes.</li> <li>▪ Heat is one of the fundamental forms of energy affecting change and order of matter in our universe.</li> </ul>
<b>Essential Questions</b>	<ul style="list-style-type: none"> <li>▪ What are important chemical and physical properties of hydrocarbons?</li> <li>▪ Why are hydrocarbons commonly used as fuels?</li> <li>▪ Why are carbon-based molecules so versatile as chemical building blocks?</li> <li>▪ What properties are important to consider in finding substitutes for petroleum?</li> </ul>
<b>Essential Knowledge</b>	<ul style="list-style-type: none"> <li>▪ Petroleum (crude oil), is a nonrenewable resource that must be refined prior to use, consists of a complex mixture of hydrocarbon molecules.</li> <li>▪ The distribution of crude oil reserves does not necessarily correspond to areas of high petroleum use.</li> <li>▪ Liquid substances can often be separated according to their differing boiling points in a process called distillation.</li> <li>▪ The atoms in hydrocarbons and in other molecules are held together by covalent bonds.</li> <li>▪ Molecules can be represented by Lewis dot structures, structural formulas, or molecular formulas.</li> </ul>
<b>Vocabulary</b>	<ul style="list-style-type: none"> <li>▪ <u>Terms:</u> <ul style="list-style-type: none"> <li>○ crude oil, hydrocarbons, distillation, distillate, fractions, fractional distillation, intermolecular forces, organic chemistry, carbon chain, electron shells, valence electrons, covalent bonds, electron-dot formulas, Lewis dot structures, structural formula, alkanes, tetrahedron, molecular formula, condensed formula, structural isomers, fossil fuels, potential energy, kinetic energy, chemical energy, thermal energy, endothermic, exothermic, specific heat, heat of combustion, cracking, catalyst, octane rating, oxygenated fuels, polymers, petrochemicals, saturated hydrocarbons, alkenes, alcohols, biomolecules, biodiesel, oil shale, compressed natural gas, hybrid vehicles, and fuel cells.</li> </ul> </li> </ul>
<b>Essential Skills</b>	<ul style="list-style-type: none"> <li>▪ Determine hydrocarbon boiling points.</li> <li>▪ Plot trends in alkane boiling points.</li> <li>▪ Calculate boiling points of isomers.</li> <li>▪ Diagram automobile energy conversions.</li> <li>▪ Determine energy efficiency and heats of combustion.</li> <li>▪ Describe fuel sources used over the years.</li> </ul>

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<p><b>Related Maine Learning Results</b></p>	<p><u>Science and Technology</u> A. Unifying Themes A1. Systems Students apply an understanding of systems to explain and analyze man-made and natural phenomena. a. Analyze a system using the principles of boundaries, subsystems, inputs, outputs, feedback, or the system's relation to other systems and design solutions to a system problem. b. Explain and provide examples that illustrate how it may not always be possible to predict the impact of changing some part of a man-made or natural system. A2. Models Students evaluate the effectiveness of a model by comparing its predictions to actual observations from the physical setting, the living environment, and the technological world. A3. Constancy and Change Students identify and analyze examples of constancy and change that result from varying types and rates of change in physical, biological, and technical systems with and without counterbalances.</p>
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<p><b>Related Maine Learning Results</b></p>	<p>B. The Skills and Traits of Scientific Inquiry and Technological Design</p> <p>B1.The Skills and Traits of Scientific Inquiry</p> <p>Students methodically plan, conduct, analyze data from, and communicate results of in-depth scientific investigations, including experiments guided by a testable hypothesis.</p> <ol style="list-style-type: none"><li>Identify questions, concepts, and testable hypotheses that guide scientific investigations.</li><li>Design and safely conduct methodical scientific investigations, including experiments with controls.</li><li>Use statistics to summarize, describe, analyze, and interpret results.</li><li>Formulate and revise scientific investigations using logic and evidence.</li><li>Use a variety of tools and technologies to improve investigations and communications.</li><li>Recognize and analyze alternative explanations and models using scientific criteria.</li><li>Communicate and defend scientific ideas.</li></ol> <p>C. The Scientific and Technological Enterprise</p> <p>C1.Understandings of Inquiry</p> <p>Students describe key aspects of scientific investigations: that they are guided by scientific principles and knowledge, and that they are performed to test ideas, and that they are communicated and defended publicly.</p> <ol style="list-style-type: none"><li>Describe how hypotheses and past and present knowledge guide and influence scientific investigations.</li><li>Describe how scientists defend their evidence and explanations using logical arguments and verifiable results.</li></ol> <p>C3.Science, Technology, and Society</p> <p>Students describe the role of science and technology in creating and solving contemporary issues and challenges.</p> <ol style="list-style-type: none"><li>Explain how science and technology influence the carrying capacity and sustainability of the planet.</li><li>Explain how ethical, societal, political, economic, and cultural factors influence personal health, safety, and the quality of the environment.</li><li>Explain how ethical, societal, political, economic, religious, and cultural factors influence the development and use of science and technology.</li></ol>
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<p style="text-align: center;"><b>Related Maine Learning Results</b></p>	<p>D. The Physical Setting D3.Matter and Energy</p> <p>Students describe the structure, behavior, and interactions of matter at the atomic level and the relationships between matter and energy.</p> <ol style="list-style-type: none"> <li>a. Describe the structure of atoms in terms of neutrons, protons, and electrons and the role of the atomic structure in determining chemical properties.</li> <li>b. Describe how the number and arrangement of atoms in a molecule determine a molecule's properties, including the types of bonds it makes with other molecules and its mass, and apply this to predictions about chemical reactions.</li> <li>c. Explain the essential roles of carbon and water in life processes.</li> <li>d. Describe how light is emitted and absorbed by atoms' changing energy levels, and how the results can be used to identify a substance.</li> <li>e. Describe factors that affect the rate of chemical reactions (including concentration, pressure, temperature, and the presence of molecules that encourage interaction with other molecules.</li> <li>f. Apply an understanding of the factors that affect the rate of chemical reaction to predictions about the rate of chemical reactions.</li> <li>g. Describe nuclear reactions, including fusion and fission, and the energy they release.</li> <li>h. Describe the radioactive decay and half-life.</li> <li>i. Explain the relationship between kinetic and potential energy and apply the knowledge to solve problems.</li> <li>j. Describe how in energy transformations the total amount of energy remains the same, but because of inefficiencies (heat, sound, and vibration) useful energy is often lost through radiation or conduction.</li> <li>k. Apply an understanding of energy transformations to solve problems.</li> <li>l. Describe the relationship among heat, temperature, and pressure in terms of the actions of atoms, molecules, and ions.</li> </ol>
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<p><b>Related Maine Learning Results</b></p>	<p>E. The Living Environment E2.Ecosystems Students describe and analyze the interactions, cycles, and factors that affect short-term and long-term ecosystem stability and change.</p> <ol style="list-style-type: none"> <li>Explain why ecosystems can be reasonably stable over hundreds or thousands of years, even though populations may fluctuate.</li> <li>Describe dynamic equilibrium in ecosystems and factors that can, in the long run, lead to change in the normal pattern of cyclic fluctuations and apply that knowledge to actual situations.</li> <li>Explain the concept of carrying capacity and list factors that determine the amount of life that any environment can support.</li> <li>Describe the critical role of photosynthesis and how energy and the chemical elements that make up molecules are transformed in ecosystems and obey basic conservation laws.</li> </ol>
<p><b>Sample Lessons And Activities</b></p>	<ul style="list-style-type: none"> <li>▪ Perform a laboratory experiment that uses distillation to separate a mixture of two liquids.</li> <li>▪ Using data of hydrocarbon boiling points make predictions by answering questions regarding many selected hydrocarbons.</li> <li>▪ In an investigation of alkanes, assemble models of several simple hydrocarbon molecules and then answer the accompanying questions.</li> <li>▪ Explore trends in alkane boiling points by plotting trends on a graph and extrapolate the results.</li> <li>▪ As an end of unit project create and present an advertisement featuring an imaginary but plausible vehicle that uses a particular type of fuel.</li> </ul>
<p><b>Sample Classroom Assessment Methods</b></p>	<ul style="list-style-type: none"> <li>▪ Sections A, B, C, and D Quizzes followed by tests after each section</li> <li>▪ Summary Questions for each section</li> <li>▪ Laboratory experiments for each section</li> <li>▪ Skill problems for various parts of each section</li> </ul>
<p><b>Sample Resources</b></p>	<ul style="list-style-type: none"> <li>▪ <u>Publications:</u> <ul style="list-style-type: none"> <li>○ <u>Chemistry in the Community</u>, Chemcom, 5<sup>th</sup> edition textbook and ancillaries</li> </ul> </li> <li>▪ <u>Videos:</u> <ul style="list-style-type: none"> <li>○ <u>World of Chemistry</u> series</li> <li>○ <u>Planet Earth</u> series</li> </ul> </li> </ul>